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T2 = final temperature (K). Chemistry:

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atm. 375 K. 1.100 atm. 350. K. 5. 0.980

atm. 30.0°C. 185 mL. 0.900 atm. 28.0°C.

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by using the combined gas law to find the
unknown variable in each row. Initial
Final volume pressure temperature volume
pressure temperature 1. 80.0 mL 96 kPa
27°C 99.0 kPa 0.0°C

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Period _____ Phase Changes Heating a

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Answers causes the temperature to increase. This causes the particles to move faster and collide harder. Harder, faster collisions cause particles to rebound harder moving them further apart. ...

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